

Advanced Solid Sorbents and Process Designs for Post-Combustion CO₂ Capture

RTI International

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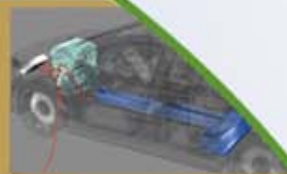
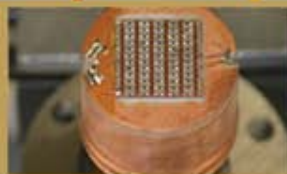
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PENNSTATE



Project Overview





Overall objective: to develop an advanced solid sorbent CO₂ capture technology based on novel, amine-based (PEI) sorbent and demonstrate a potential to achieve DOE's CO₂ capture targets of >90% CO₂ capture with <35% increase in cost of electricity

Project Details

- Combination of two previously funded DOE projects
- Project Cost: \$3,847,161
 - DOE Share: \$2,997,038
 - Cost Share: \$850,123
- Period of performance: 10/1/2011 to 9/30/2014
- Prime Recipient: RTI International

Project Objectives

- Improve stability and performance of novel amine-based (PEI) “**Molecular Basket Sorbent**” and make sorbents applicable for RTI's **fluidized, moving-bed reactor design (FMBR)**
- Improve reactor design; optimize operability & heat integration
- Conduct prototype testing and prove that the technology reduces parasitic energy load and capital and operating costs associated with CO₂ capture (economic study)

				
<ul style="list-style-type: none"> ▪ Project management ▪ Process design ▪ Fluidized-bed sorbent 	<ul style="list-style-type: none"> ▪ PSU's EMS Energy Inst ▪ PEI and sorbent improvement 	<ul style="list-style-type: none"> ▪ Masdar Carbon ▪ Masdar Institute ▪ NGCC application 	<ul style="list-style-type: none"> ▪ Techno-economic evaluation ▪ Process design support 	<ul style="list-style-type: none"> ▪ Sorbent scale-up ▪ Commercial manufacture evaluation

RTI's Center for Energy Technology (CET)



RTI's Johnson Science and Engineering Building
Home of RTI's Center for Energy Technology

- RTI International was established in 1958 in RTP, North Carolina
- One of the world's leading research institutes
- **Mission:** To improve the human condition by turning knowledge into practice
- CET develops **advanced energy technologies** to address some of the world's great energy challenges

Advanced Gasification

- Syngas cleanup/conditioning
- Substitute natural gas production

Biomass & Biofuels

- Biomass gasification
- Pyrolysis to biocrude and conventional fuels

CO₂ Capture & Conversion

- Post-combustion CO₂ capture
- Pre-combustion CO₂ capture
- CO₂ conversion

Fuels & Chemicals

- Syngas conversion
- Hydrocarbon desulfurization

Water & Energy

- Industrial water reuse
- Energy and waste heat recovery

Shale Gas

- Gas separation & processing
- Process water treatment





Project Roles

- Project co-funding partner
- Providing engineering expertise to evaluate technology feasibility and potential application to natural gas-fired plants
- R&D efforts focused on parallel development of technology for natural gas application

Memorandum of Understanding

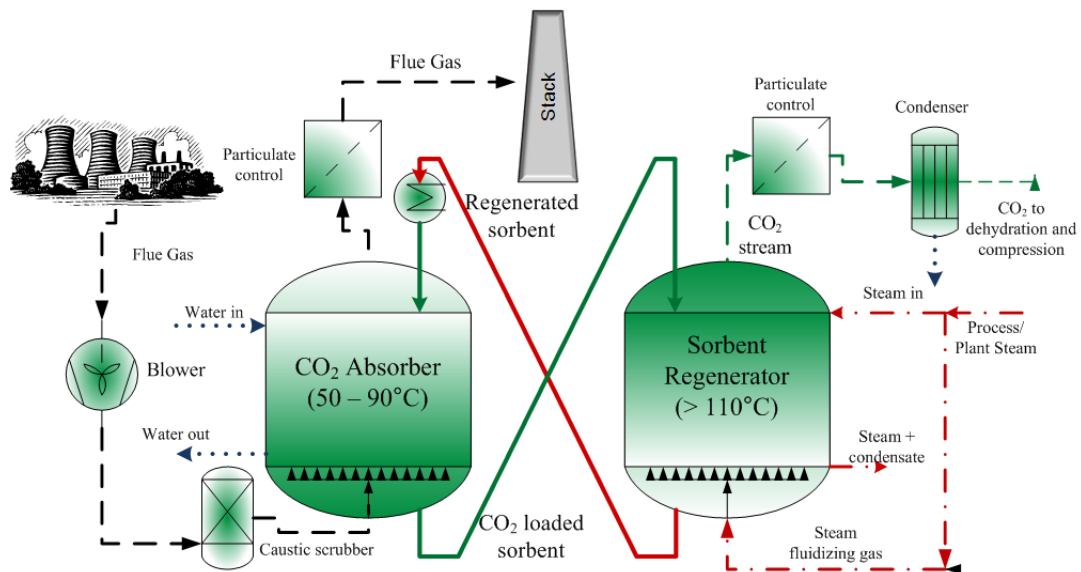
- DOE & Masdar signed MoU in 2010 to promote collaboration on clean/sustainable energy technologies
- Cooperation in three key areas: Carbon Capture & Sequestration; Water & Biofuels; Building Technology

Masdar

Masdar Overview

- Established in 2006, based in Abu Dhabi, United Arab Emirates
- Commercially-driven enterprise that aims to reach the broad boundaries of the renewable energy and sustainable technology industry and help drive the development of clean fossil fuel energy technologies
- Operates through five integrated units:
 - Masdar Carbon • Masdar Institute • Masdar Capital • Masdar Power • Masdar City
- **Masdar Carbon's** flagship initiative involves the development of a large-scale CCUS network in Abu Dhabi
- **Masdar Institute** is an independent, research-driven graduate university, partnered with MIT

Engineering a Sorbent-based CO₂ Capture Process



Challenges

- Heat management is still critical; need for heat integration
- Water management is critical to avoid condensation
- Pressure drop needs to be controlled
- Process design has not been evaluated experimentally

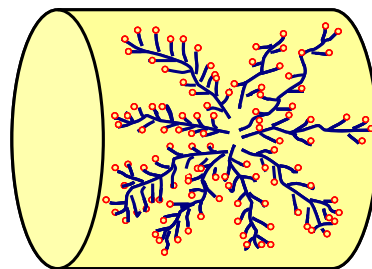
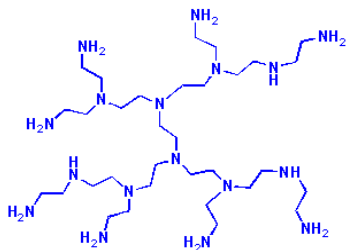
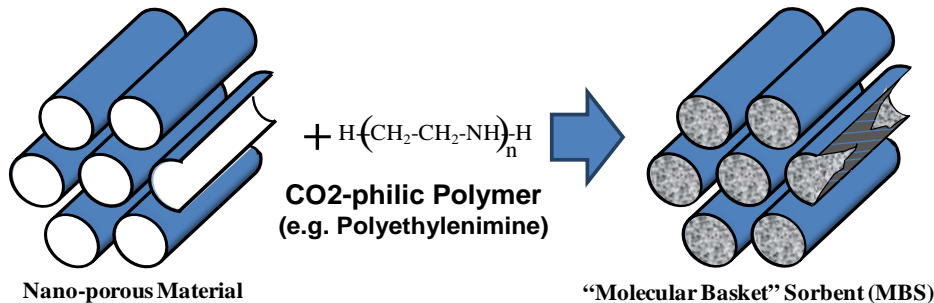
Notes from engineering study / experience in field

- Sorbent working capacity must be sufficient (8+ wt% CO₂) to control sorbent transport demands and scale of process
- Heat generated during capture leads to large temperature excursions in Absorber and reduced capture efficiency
- Sorbents have low C_p unlike aq. amine solvents, this internal heat management systems are necessary
- Commercial-scale fixed-bed processes have inadequate heat removal rates

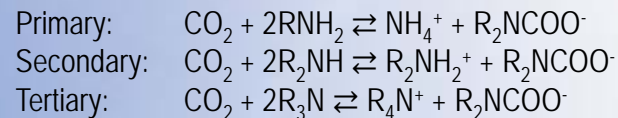
RTI's Fluidized, Moving-bed Reactor process concept:

- Minimizes # of process vessels and sorbent load
- Superior gas-solid heat and mass transfer characteristics
- Minimizes thermal regeneration energy
- Reduced pressure drop once bed is fluidized
- Counter current gas-solids flow maximizes CO₂ driving force
- Minimizes sorbent attrition by avoiding entrained-bed mode
- Commercial embodiments of equipment

PSU's MBS CO₂ Absorbing Material



Polymeric amines have high density of CO₂ absorbing sites. Sorbent carbamate chemistry:



PSU has been working in field for ~10 years
(Dr. Chunshan Song's research group)

PSU's Molecular Basket Sorbent (MBS) material is one of the most promising impregnated amine sorbents:

- High CO₂ loadings (>14 wt% CO₂)
- High CO₂/N₂ selectivity (> 1000)
- Reasonable heat of absorption (66 kJ/mol)
- Utilizes CO₂-philic polymer, polyethyleneimine (PEI)

Challenges

- Poor performance stability above 90°C
- Currently in powdered or fixed-bed form
- Too costly (uses mesoporous silica supports)

Sorbent Stability

- Explore cross-linking agents
- Different PEIs and strengthening PEI-support bond
- Support surface modification

Sorbent Form

- Convert existing MBS platform to a fluidizable and attrition-resistant material
- Explore impregnation, spray drying, and binders

Sorbent Cost

- Utilize low-cost, commercially-available supports
- Work with Süd-Chemie to devise low-cost commercial manufacturing techniques

Design Effectiveness

- Explore commercial embodiments of equipment
- Kinetic and equilibrium testing for detailed design
- Detailed CFD modeling

Economics

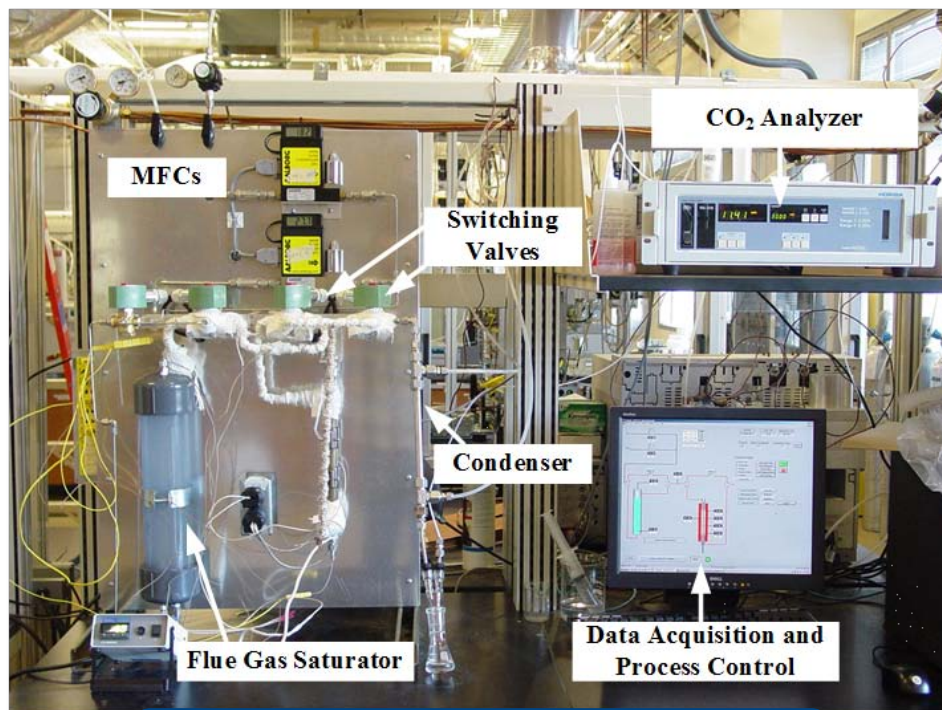
- True economic potential of combined (MBS + FMBR design) has not yet been studied
- Conduct rigorous techno-economic evaluation

Project Schedule, Work Plan, Milestones

	10/1/11	12/31	3/31/12	6/30	9/30
Task 1: Project Management and Planning	[Blue bar]				
Task 2: Preliminary Technology Feasibility Study	[Blue bar]				
Task 3: PEI Improvement and Fluidized-bed Sorbent Development	[Blue bar]				
Task 4: Fluidized-bed Modeling and Contactor Design	[Blue bar]				

	Milestone / Criteria	Status
★1	<ul style="list-style-type: none"> ▪ Complete preliminary technology feasibility study ▪ Results improve upon SOTA and show potential to meet DOE targets 	Complete
★2	<ul style="list-style-type: none"> ▪ Successful scale-up of fluidized-bed MBS material ▪ 15 kg of sorbent produced having same properties as lab-scale 	Delayed, but in-progress
★3	<ul style="list-style-type: none"> ▪ Working multi-physics, CFD model of FMBR design ▪ Elicit effects of process elements on CO₂ capture, heat/mass transfer 	In-progress
★4	<ul style="list-style-type: none"> ▪ Fabrication-ready design package for single-stage FMBR unit ▪ Complete engineering drawings, cost estimate, and schedule 	On schedule

Testing Equipment



RTI's Fully Automated, Packed-Bed Reactor System for Multi-cycle CO₂ Capture-Regeneration Testing

Objective: Improve the thermal and chemical stability of base PEI reactant while transitioning fixed-bed MBS material into a fluidizable form

RTI's automated packed-bed reactor systems

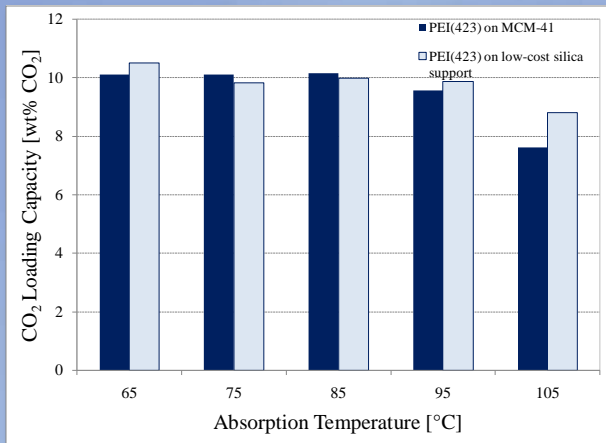
Capabilities

- Fully-automated operation, data analysis
- Multi-cycle, absorption-regeneration experiments
- Rapid sorbent screening experiments
- Comprehensive parametric evaluation
- Long-term effect of contaminants

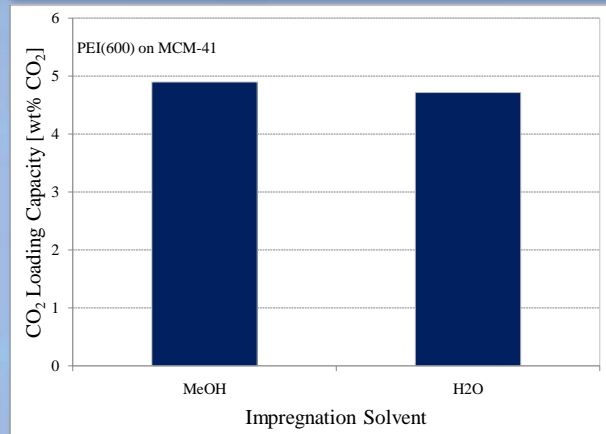
Measureables

- Dynamic CO₂ loading capacity
- Rate of CO₂ loading on sorbent (wt%/min)
- Thermal waves due to absorption or desorption
- Pressure drop across packed bed
- PBR #1 – sorbent screening
- PBR #2 – multi-cycle stability testing

Towards producing a low-cost, attrition-resistant sorbent



RTI's lab-scale spray dryer



RTI's Davison Cup Attrition Tester

Motivation

- Need new support that is low-cost, fluidizable & attrition-resistant
- Need preparation method that is feasible for commercial scale-up

Methodology

- Screened 20+ commercially-available, low-cost support materials, Tested with automated, packed-bed absorption system
- Modified preparation procedure

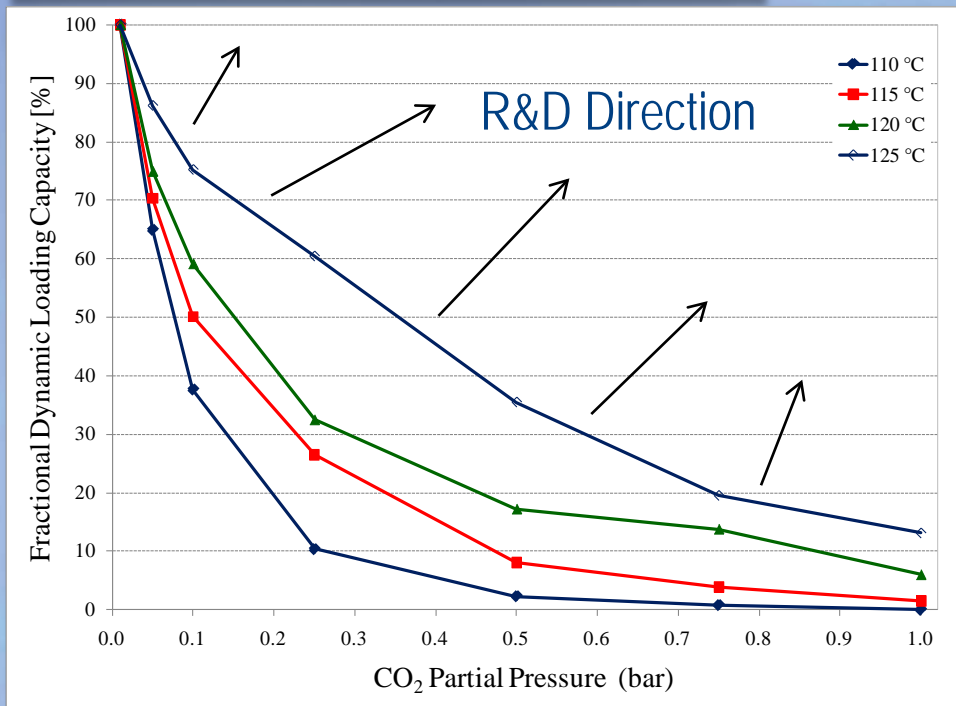
Results

- Identified a low-cost support material yielding a sorbent with comparable performance to expensive, mesoporous silica
 - *Cost of MCM-41: on the order of \$1,000/kg*
 - *Cost of low-cost alternative: < \$5/kg*
- Low-cost support is also more amenable to forming a strong, attrition-resistant, fluidizable particle
- Demonstrated that water can replace methanol (or any alcohol) as the PEI impregnation solvent

Conditions	Absorption	Regeneration
Temperature [°C]	65	110-125
Gas Composition [vol%]	13.3% CO ₂ , 5% H ₂ O, 2.35% O ₂ , bal. N ₂	0-100% CO ₂ , 0-5% H ₂ O, bal. N ₂
GHSV [h ⁻¹]	3,000	3,000
Fresh sorbent for each regeneration temperature		

Regeneration Under Realistic Conditions

Evaluated effect of regeneration temperature and CO₂ partial pressure on dynamic loading capacity



Motivation

- Need to avoid, to the greatest extent possible, high concentration steam sweep or vacuum regeneration

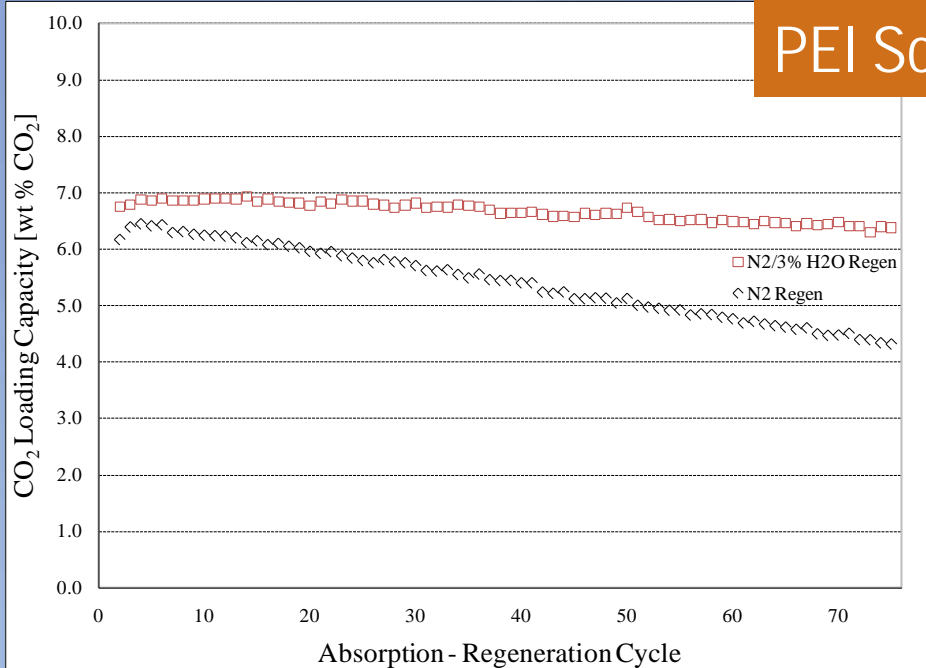
Methodology

- Automated, packed-bed absorption system
- Sorbent cycled between abs. and regen. conditions with increasing CO₂ content in the regen. gas
- Fresh sorbent sample used for each regen. Temperature

Results

- Increasing P_{CO₂} has a significant detrimental effect on the dynamic loading capacity of the sorbent
- Increasing the regen. temperature drastically improves the dynamic loading capacity

PEI Sorbent Stability – Water



Determine the effect of moisture in regeneration gas on long-term performance stability

Motivation

- Explore operating conditions to improve sorbent stability

Methodology

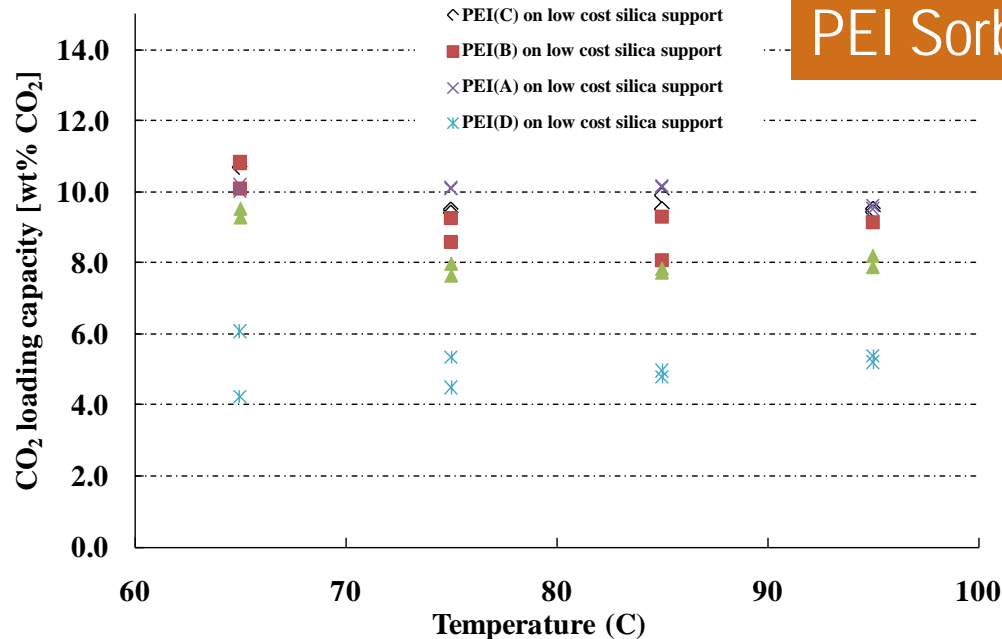
- Automated, packed-bed absorption system
- 75 abs. and regen. cycles at prescribed conditions

Results

- Addition of a small amount of moisture to the regen. gas dramatically improves the multi-cycle performance stability
- Similar findings to the Sayari (UoF) and Jones (GIT) groups
- Improvement most likely related to reducing the formation of thermally-stable urea under regeneration condition
 - $2\text{RNH}_2 + \text{CO}_2 \leftrightarrow \text{RNH-CO-NHR} + \text{H}_2\text{O}$
- Observed reduction in carbonyl peak for sample regenerated in H₂O/N₂ using DRIFTS

Conditions	Absorption	Regeneration
Temperature [C]	75	105
Gas Composition [vol%]	13.3% CO ₂ , 5% H ₂ O, 2.35% O ₂ , bal. N ₂	a. 3% H ₂ O, bal N ₂ b. 100% N ₂
GHSV [h ⁻¹]	3,000	3,000
Fresh sorbent was used for each experiment		

PEI Sorbent Performance – PEI Type



Determine the effect of different PEI types on sorbent CO₂ capture performance

Motivation

- Baseline PEI(423) may not be suitable for commercial use

Methodology

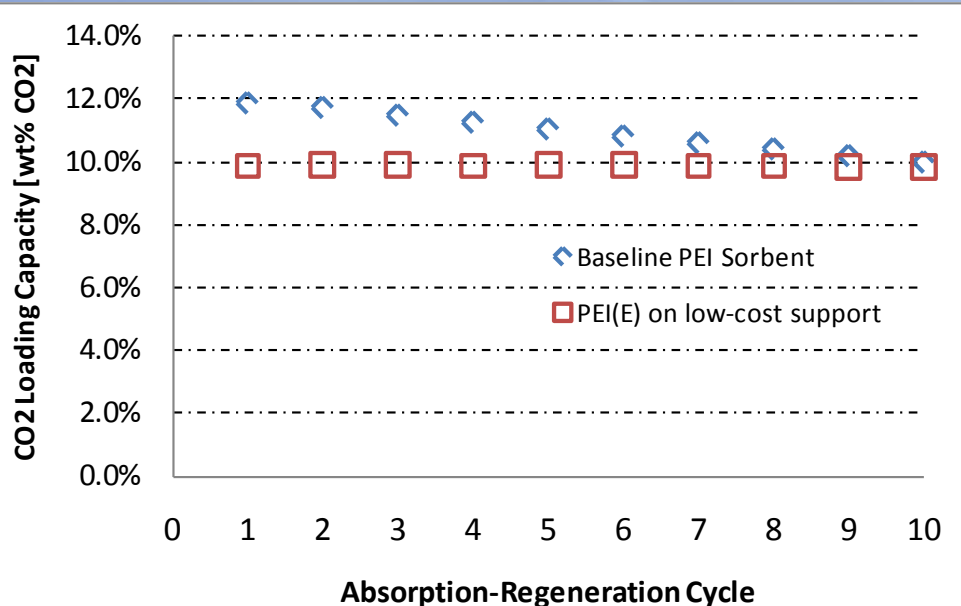
- Automated, packed-bed absorption system
- Single abs. and regen. cycle at different temperatures

Results

- Sorbent CO₂ capture performance can vary widely as a result of using different PEI types
- Identified most promising PEI types for further screening

Conditions	Absorption	Regeneration
Temperature [C]	Varied	110
Gas Composition [vol%]	14.8% CO ₂ , 5.7% H ₂ O, 2.61% O ₂ , bal. N ₂	2.5% H ₂ O, bal N ₂
GHSV [h ⁻¹]	3,000	3,000

PEI Sorbent Stability – PEI Type



Evaluate a promising new PEI type for improved sorbent thermal stability

Motivation

- Alternative PEIs and preparation procedure must lead to improved sorbent stability

Methodology

- Absorption and regeneration cycling using thermogravimetric analyzer (TGA)
- Prepared new sorbent sample using PEI type – PEI(E) on low-cost silica support

Results

- Significant thermal stability improvement over 10 cycles at 120°C regeneration compared to baseline sorbent
- Slightly lower starting CO₂ capacity

Conditions	Absorption	Regeneration
Temperature [C]	90	120
Gas Composition [vol%]	100% CO ₂	100% N ₂

Preliminary Technology Feasibility Study¹

Methodology

- Detailed PFDs of combined CO₂ capture and compression systems and major process components
- AspenPlus® to develop simulations and complete stream tables (assumptions and experimental data used)
- Detailed process equipment lists, consumables, and sizing using Aspen Process Economic Analyzer (PEA)
- Equipment cost estimates, operating costs, parasitic power load, levelized COE using Aspen PEA and quotes

Highlights

- Developed detailed estimates for a) parasitic power loss, b) levelized cost of power and steam, c) itemized installed equipment costs, d) cost of CO₂ capture, e) percent decrease in plant efficiency
- Estimated **increase in cost of electricity ~ 50%** with potential to reduce further through a) reactor refinements, b) sorbent cost reduction, c) improve sorbent strength, d) heat integration strategies, e) alternative SO₂
- Compared to SOTA MEA process, a) ~ **40% less steam consumption**, b) ~ **5 point improvement** in decreased plant efficiency, c) ~ **45% improvement in cost of CO₂ capture**

Sensitivity Analyses

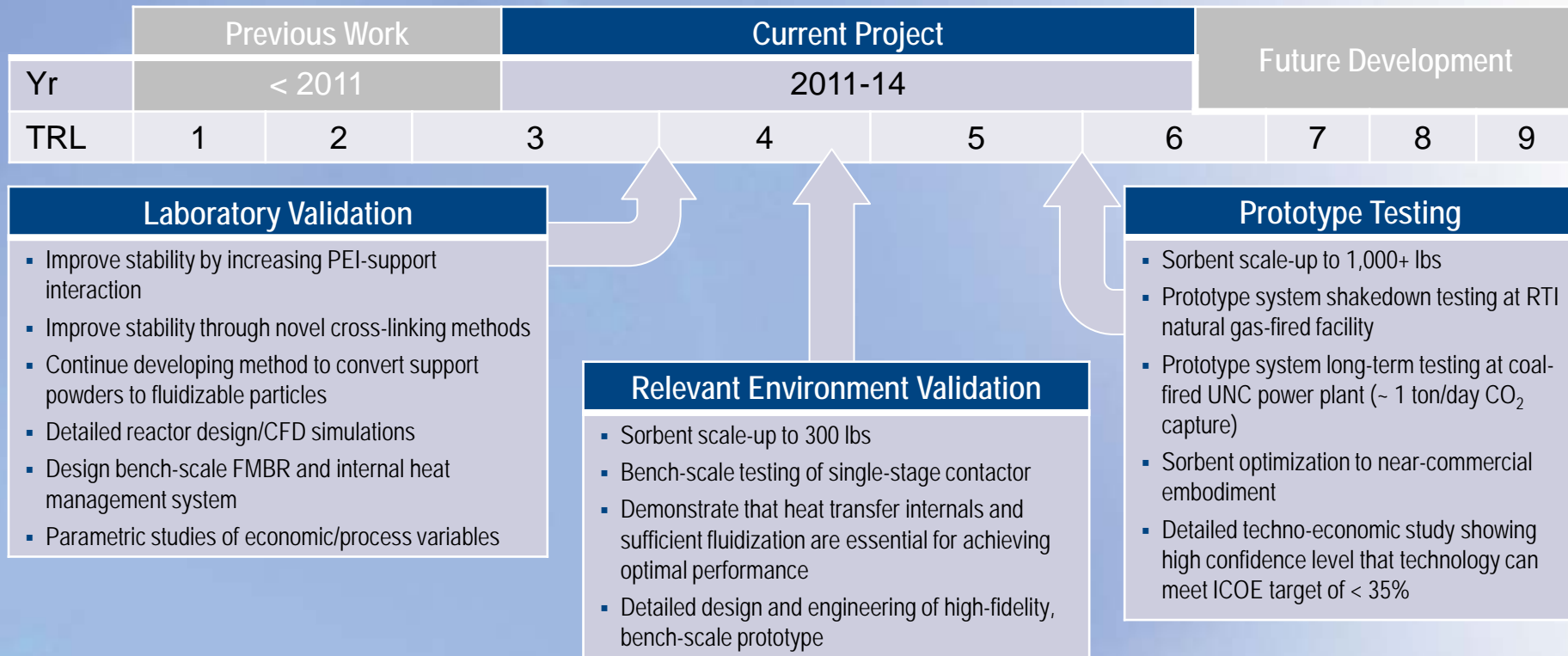
- Sorbent loss rate, price, capacity
- Alternative SO_x removal processes
- Stripping steam demand
- Reactor design based on equilibrium-limited kinetic model

R&D Directions

- Kinetic/equilibrium studies
- Long-term contaminant studies
- Study effects of particle size
- Detailed design study of FMBR

¹Basis: Attachment 3 – “Basis for Technology Feasibility Study” of the Funding Opportunity Announcement, DE-FOA-0000403

Plans for Future Development



Prototype Testing and Beyond

RTI's Central Utility Plant (CUP)



- ~ 5 MW_t boiler – Natural Gas-fired
- 13,800 lbs/hr steam production
- CO₂ produced: ~ 25 TPD

UNC–Chapel Hill Co-Generation Power Plant



- Generates steam and electricity
- Coal Feed: ~ 50,000 lbs/hr
- CO₂ Produced: 1500 TPD

Future Development

2014 - 16		2016-19	2020+
TRL-6	7	8	9

Pilot Validation (2014 – 2016)

- 5 MW (eq) pilot-scale testing at coal-fired utility
- > 100 ton/day CO₂ capture

Demonstration (2016 – 2019)

- 50 -100 MW (eq) pre-commercial scale testing

Commercial (2020+)

Prototype testing objectives

- Demonstrate long-term stability and performance of MBS-based sorbent and FMBR process with actual coal-fired flue gas
- Shakedown and troubleshoot at RTI
- Parametric and long-term testing (1,000 hrs) at UNC's power plant. Sorbent scale-up

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RTI	PSU	Masdar Carbon	Masdar Institute	DOE/NETL
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